

Binary Ionic Compounds Binary

Name each of the following compounds:

Formula	Name	Formula	Name
1. CuS		11. CuCl ₂	
2. PbBr ₄		12. CuBr	
3. Pb ₃ N ₂		13. PbO	
4. Fe ₂ O ₃		14. Fe ₂ S ₃	
5. FeI ₂		15. PbCl ₂	
6. Sn ₃ P ₄		16. SnO	
7. Cu ₂ S		17. Cu ₂ O	
8. SnCl ₂		18. PbO ₂	
9. HgO		19. FeO	
10. Hg ₂ F ₂		20. SnO ₂	

Write the formula for each of the following compounds:

Name	Formula	Name	Formula
1. mercury(I) oxide		11. lead(II) oxide	
2. mercury(I) iodide		12. iron(III) bromide	
3. gold(III) chloride		13. copper(II) oxide	
4. manganese(II) oxide		14. mercury(I) selenide	
5. chromium(III) chloride		15. lead(IV) chloride	
6. cobalt(II) oxide		16. copper(II) nitride	
7. manganese(III) oxide		17. tin(IV) chloride	
8. cobalt(III) sulphide		18. tin(IV) oxide	
9. gold(I) fluoride		19. iron(III) sulphide	
10. chromium(II) bromide		20. iron(II) sulphide	