

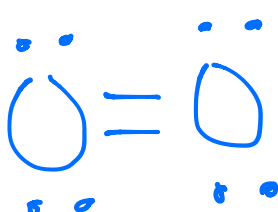

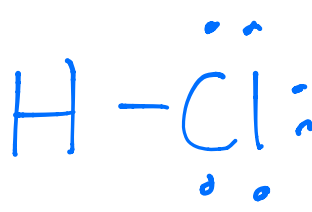
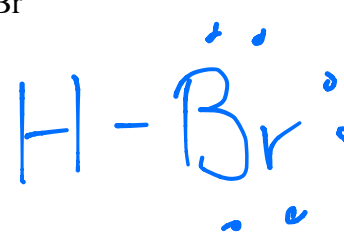

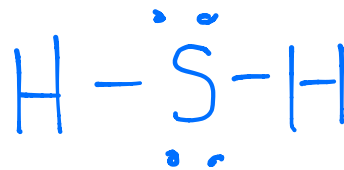
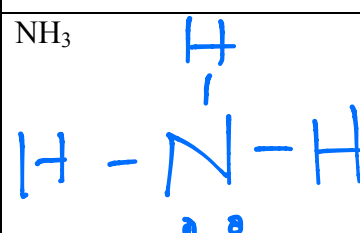
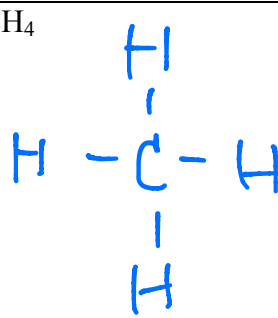
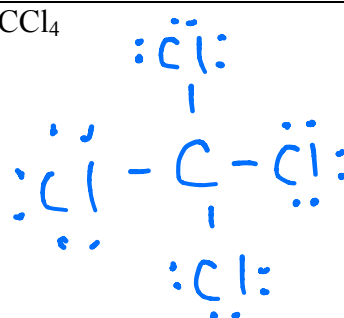
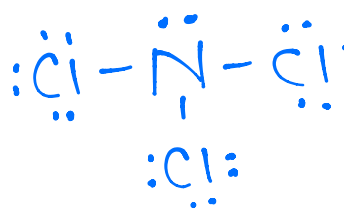
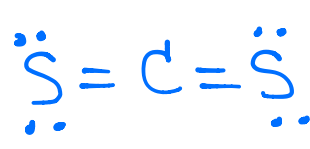
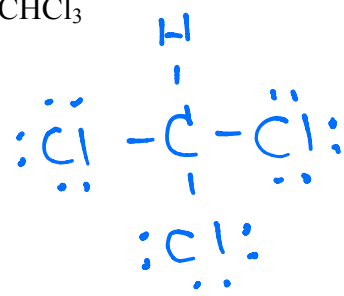


1. Draw Lewis structures for the following **covalent** compounds:

H <sub>2</sub> 	F <sub>2</sub> 	O <sub>2</sub> 
N <sub>2</sub> 	HCl 	HBr 
H <sub>2</sub> O 	H <sub>2</sub> S 	NH <sub>3</sub> 
CH <sub>4</sub> 	CCl <sub>4</sub> 	NCl <sub>3</sub> 
CS <sub>2</sub> 	CHCl <sub>3</sub> 	HCN 