## **Chemical Equations Worksheet**

Write the skeleton equations for each of the following chemical reactions:

1)	When aqueous beryllium chloride reacts with aqueous silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made.
2)	When isopropanol ( $C_3H_8O$ ) burns in oxygen, carbon dioxide, water, and heat are produced.
3)	When aqueous sodium hydroxide reacts with sulfuric acid ( $H_2SO_4$ ), aqueous sodium sulfate, water, and heat are formed.
4)	When fluorine gas is put into contact with calcium metal at high temperatures, calcium fluoride powder is created in an exothermic reaction.
5)	When sodium metal reacts with iron (II) chloride, iron metal and sodium chloride are formed.

Write word equations for the following skeleton equations.

1) Al 
$$(s) + O_2(g) \rightarrow Al_2O_3(s)$$

2) 
$$C_2H_{6(g)} + O_{2(g)} \rightarrow CO_{2(g)} + H_2O_{(g)}$$

3) 
$$Fe_2O_3$$
 (s) +  $H_2SO_4$  (aq)  $\rightarrow$   $Fe_2(SO_4)_3$  (aq) +  $H_2O$  (l)

4) 
$$CO_{2(g)} + H_{2(g)} \rightarrow CH_{4(g)} + H_2O_{(g)}$$