

Ionic Compounds Explanation Assignment

Show how to write formulas & names of ionic compounds.

 Write the **formula** for <u>ONE</u> of the following compounds. <u>Show</u> all the steps required to write the formula like on the worksheets.

<u>DO NOT</u> say "criss-cross" as an explanation.

- iron (III) sulfide nickel (III) nitride copper (I) oxide
- 2) Show all the steps required to find the **name** for **ONE** of the following compounds like on the worksheets.
 - Sn(OH)₂ AuPO₄ Pb(ClO₃)₂

Rubric:

 = unclear / several errors 	② = complete / mostly correct	③= thorough / meticulous
--	-------------------------------	--------------------------

	Clarity of explanation (visually & verbally)	Proper Terminology
		(1)= rarely
Formula	1 2 3	②= sometimes
		③=often
Naming	1 2 3	
		⑤= always

Sample terminology:

ie. Cation, anion, subscript, charge, formula, metal, non-metal, /11 transition metal, group number, multivalent, polyatomic, binary....