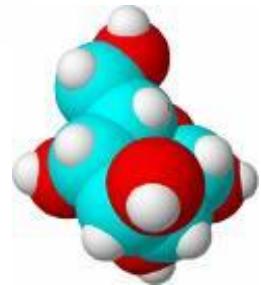
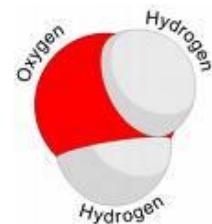
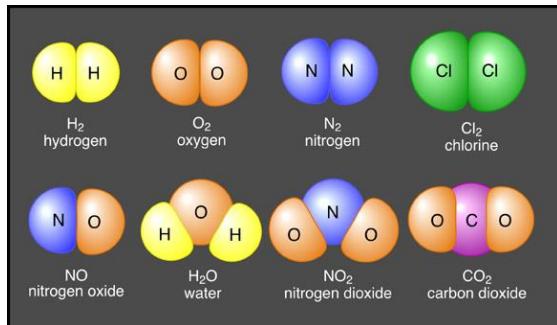


## Topic 4: Compounds

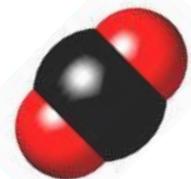
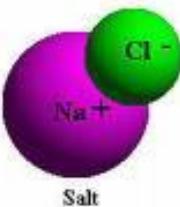
### Compounds

- Made up of \_\_\_\_\_ or more \_\_\_\_\_
- \_\_\_\_\_ be broken down into the elements that make them up
- Are \_\_\_\_\_ substances
- Elements combine in set \_\_\_\_\_ (formulas)



### What is a formula?

- Formulas tell us how many \_\_\_\_\_ there are of each \_\_\_\_\_
- The formula for table salt is \_\_\_\_\_
  - Na is the symbol for \_\_\_\_\_
  - Cl is the symbol for \_\_\_\_\_
  - There are \_\_\_\_\_ atoms total
- The formula for carbon dioxide is \_\_\_\_\_
  - C is the symbol for \_\_\_\_\_
  - O is the symbol for \_\_\_\_\_
  - There are \_\_\_\_\_ atoms total



Name of Compound	Formula	# of different elements	Names of elements and how many atoms of each		Total # of atoms
calcium chloride	CaCl <sub>2</sub>				
iron sulfate	FeSO <sub>4</sub>				
aluminum oxide	Al <sub>2</sub> O <sub>3</sub>				