Ionic



Atom



Ion



Covalent



Polyatomic



Binary



Multivalent



Anion



Cation



Electron



**SOLO Hexagons**

**Unistructural LO** - single hexagon.

**Multistructural LO** - several separate hexagons.

**Relational LO** - connected hexagons (explain the cause for connecting two edges).

**Extended abstract** - tessellated hexagons (generalise about the vertex where three hexagons meet).

Proton



Acid



Base



Conservation of Mass



Subscript



Coefficient



Neutralization



Reaction



Chemical change



Single Displacement



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Decomposition



Combustion



Synthesis



Indicator



Reactant



Product



Group



Valence shell



Neutron



Period



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