

Balancing Chemical Equations Individual Puzzle 1

$$\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$$

$$\text{Al} + \text{HCl} \rightarrow \text{AlCl}_3 + \text{H}_2$$

$$\text{S} + \text{O}_2 \rightarrow \text{SO}_2$$

$$\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$$

$$\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$$

$$\text{C}_2\text{H}_2 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$

$$\text{Cr} + \text{O}_2 \rightarrow \text{Cr}_2\text{O}_3$$

$$\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$$

$$\text{AlBr}_3 + \text{Cl}_2 \rightarrow \text{AlCl}_3 + \text{Br}_2$$

$$\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$

$$\text{AgNO}_3 + \text{K}_3\text{PO}_4 \rightarrow \text{Ag}_3\text{PO}_4 + \text{KNO}_3$$

$$\text{NH}_3 + \text{N}_2 \rightarrow \text{H}_2$$

$$\text{Ag} + \text{O}_2 \rightarrow \text{Ag}_2\text{O}$$

$$\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$

$$\text{Ca}(\text{OH})_2 + \text{H}_2\text{O} \rightarrow \text{HPO}_4$$

$$\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$