


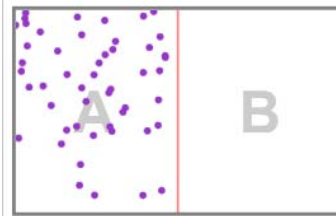
Student Exploration: Diffusion

Activity A: Temperature and diffusion	Get the Gizmo ready: Set the Wall to 100%.	Wall  100 %
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Question: How does temperature affect the rate of diffusion?

1. Set the temperature (**Temp.**) to 100 K, press **Play**.
2. Observe the motion of particles.
3. Click **Reset**.

Temp.  100 K Controls: 



4. Set the temperature to 600 K, click **Play**, and observe.

Temp.  600 K 

Describe how the particles move when the temperature is higher _____

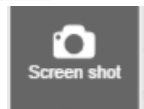
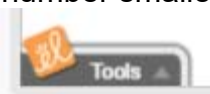
(faster/slower, more/less...)

Find the time to reach equilibrium for each temperature:

5. Set the temperature (**Temp.**) to 100 K, press **Play**.
6. Let the molecules move until they look spread out.
7. Click the TABLE tab
8. Look through the X in A until you see a number smaller than 55%
9. Record the time in the chart below.
10. Click the Tools Tab, select screen shot.

DESCRIPTION TABLE BAR CHART GRAPH

Time (s)	x in A (%)	y in B (%)
39	54.00	0
40	48.00	0
41	50.00	0
42	50.00	0
43	50.00	0
44	44.00	0



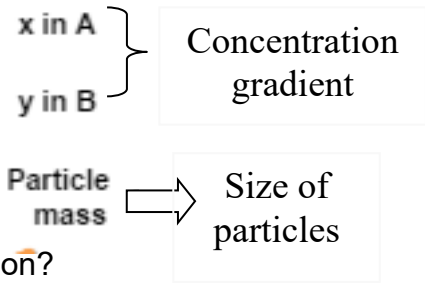
10. Set the temperature to 600 K, click **Play**,
11. Repeat steps 6-10

Time to reach equilibrium

Temperature	Time (s)
100K	
600K	



Question: How do factors other than temperature affect the rate of diffusion?



1. Click **Reset**
2. Pick a variable to investigate.
Which one did you choose? _____
3. Form hypothesis: How do you think this variable will affect rates of diffusion?
(If...then... because...) _____

In an **experiment**, only one variable is changed.

4. In the list below **circle** the one variable you picked in #1 that will be changing.
5. Set up your experiment so that that only difference is in the variable you circled. Everything else needs to stay the same.
6. List the settings you will use for each set-up below.

Set-up 1

Wall _____

x in A _____

y in B _____

Temp. _____

Particle mass _____

Set-up 2

Wall _____

x in A _____

y in B _____

Temp. _____

Particle mass _____

7. Use the Gizmo to fill in each table.
The "time to reach equilibrium" is the time it takes for the number of x particles in region A to reach **55% or lower**.

Time (s)	x in A (%)	y in B (%)
45	56.52	54.55
46	56.52	54.55
47	52.17	54.55
48	47.83	54.55
49	47.83	57.58

Set-up 1

Trial	Time to reach equilibrium
1	
2	
3	
4	
5	

Set-up 2

Trial	Time to reach equilibrium
1	
2	
3	
4	
5	

7. Compare the times to reach equilibrium for each set-up.
How did the variable you investigated affect the rate of diffusion? _____

